

Diastereoselective Aziridination of Chiral α -Carbonyl Enoates¹

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Abstract: Nosyloxycarbamates very efficiently aziridinate optically active α -carbonyl enoates with high levels of diastereoselectivity under mild conditions and in a straightforward process.

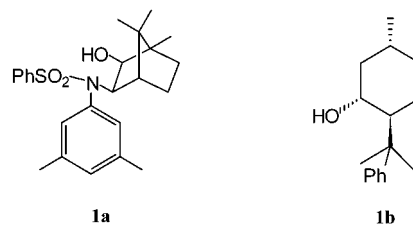
Aziridines receive continuous and increasing interest in organic synthesis² as they can possess both biological and pharmacological activities.³ Nevertheless, despite their importance, the dearth of single-step preparation methods of aziridines limits their applicative use.

Direct aziridination of alkenes by carbamates has proved to be a useful reaction.⁴ We have proposed the use of ethyl nosyloxycarbamate (NsONHCO₂Et, Ns = 4-nitrobenzenesulfonyl) in the presence of inorganic bases to obtain aziridines starting from different kinds of substituted alkenes.⁵

Recently, we reported a simple and efficient synthesis of *N*-protected aziridines, starting from 1,1-dicarbonyl-substituted olefins using either ethyl or *tert*-butyl nosyloxycarbamate in the presence of calcium oxide.⁶ Moreover, the stereoselective amination of analogous substrates by conjugate addition of hydroxylamino reagents in the presence of Evans' catalyst⁷ has been recently reported.⁸ A variation of this procedure in two steps giving aziri-

dines from the same starting materials was then reported by the same authors.⁹

Continuing our studies, we report here the aziridination of optically active α -carbonyl enoates derived from the commercially available alcohols (R*OH) Helmchen's auxiliary¹⁰ (**1a**) and (–)-8-phenylmenthol (**1b**).



All substrates were easily prepared; aziridination reactions were carried out in dichloromethane at 0 °C. Calcium oxide and ethyl or *tert*-butyl nosyloxycarbamate were added to the solution of the enoates **2–4**. The expected aziridines **5–10** were obtained after flash chromatography on silica gel, in the yields displayed in the Table 1.

No significant differences in reactivities were observed using both carbamates, and all reactions proceeded in high yields. Moreover, the aziridinations took place with a very high asymmetric induction using both chiral auxiliaries, as shown by ¹H and ¹³C NMR and HPLC analyses.¹¹

Substrates **2a** and **3a** were already reported to undergo asymmetric conjugate nucleophilic addition from the less hindered prochiral *si*-face to give 5-alkyl-substituted derivatives with a high level of diastereoselection.¹² On the basis of these results, we suggest the configuration (*S,S*) for the major diastereomers of aziridines **5a–8a**.

We also observed a very good diastereoselectivity using (–)-8-phenylmenthol as the chiral auxiliary. On the other hand, it has been reported that α,β -unsaturated esters derived from (–)-8-phenylmenthol gave other conjugate additions with a low asymmetric induction.¹³

In conclusion, we believe that this method offers considerable advantages for the development of a direct diastereoselective synthesis of chiral substituted aziridines in high yields under very mild conditions and with operational simplicity.

Experimental Section

General. GC analyses were performed on a capillary column (methyl silicone, 12.5 m \times 0.2 mm). GC MS was performed using a quadrupole mass spectrometer on a capillary column (phenyl methyl silicone, 30 m \times 0.25 mm). IR spectra were recorded on

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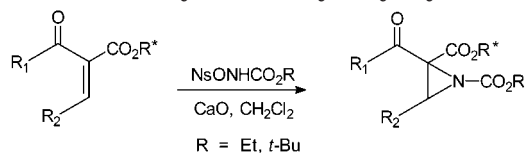
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TABLE 1. Aziridination of Optically Active α -Carbonyl Enoates by Nosyloxycarbamates

entry	substrate	time (h)	product	yield (%)	de (%)
1		2		91	99
				90	99
2		2.5		92	97
				90	98
3		2		90	99
				88	99
4		3		78	96
				79	96

a FTIR spectrophotometer in CCl_4 as the solvent. ^1H NMR and ^{13}C NMR spectra were recorded at 200 or 300 and 50 or 75 MHz, respectively. CDCl_3 was used as the solvent and CHCl_3 as the internal standard. ESI MS analyses were performed using a triple-quadrupole mass spectrometer equipped with an ESI source and a syringe pump. The experiments were conducted in the positive ion mode. HPLC analyses were performed with an instrument equipped with a differential refractometer, using an analytical column (3.9×300 mm, 80:20 hexane/ethyl acetate, flow rate 1.3 mL/min). Eluents were HPLC grade. Optical rotations were recorded at the sodium D line with a polarimeter at room temperature. NaH (55/65% suspension in mineral oil) was washed twice with pentane and dried under nitrogen.

Anhydrous THF and toluene were used as such. Substrates **2a**, **2b**, and **3a** were easily prepared following reported procedures.^{12,14}

Synthesis of 4b. According to a reported procedure,¹⁵ 8-phenylmenthyl 2-methyl-3-oxobutanoate¹⁶ (0.68 g, 2.0 mmol) was added slowly to a stirred suspension of NaH (0.07 g, 3.0 mmol)

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in 40 mL of anhydrous THF at 0 °C. After 1 h, a solution of PhSeCl (0.38 g, 2.0 mmol) in 5 mL of anhydrous THF was added dropwise. After 30 min, the solution was diluted with diethyl ether, washed with saturated NaHCO₃ and saturated NaCl, and dried over Na₂SO₄. The solvents were evaporated under reduced pressure. The crude mixture was dissolved in 20 mL of CH₂Cl₂ and stirred with 40% H₂O₂ (0.6 mL, 8 mmol) for 3 h at 0 °C. Then, the mixture was washed with saturated NaHCO₃ and saturated NaCl and dried over Na₂SO₄. After solvent evaporation, **4b** was obtained (0.63 g, yield 96%) and used without further purification: [α]_D -21.46 (c 4.8, CHCl₃); IR 1715, 1630 cm⁻¹; ¹H NMR δ 0.64–2.18 (m, 8 H), 0.93 (d, *J* = 6.6 Hz, 3 H), 1.26 (s, 3 H), 1.32 (s, 3 H), 2.38 (s, 3 H), 4.83–5.08 (m, 1 H), 5.74 (d, *J* = 1.5 Hz, 1 H), 6.15 (d, *J* = 1.5 Hz, 1 H), 7.06–7.50 (m, 5 H); ¹³C NMR δ 21.7, 24.3, 26.4, 28.5, 31.2, 31.9, 34.8, 39.7, 45.4, 54.2, 75.3, 125.2, 125.7, 128.0, 133.6, 140.9, 151.5, 163.5, 196.6; GC MS *m/z* 328 (M⁺, <1), 120 (10), 119 (100), 118 (26), 91 (23), 43 (10). Anal. Calcd for C₂₁H₂₈O₃: C, 76.79; H, 8.59. Found: C, 77.01; H, 8.47.

Typical Experimental Procedure of Aziridination. To a stirred solution of the substrate (2.0 mmol) in 5 mL of CH₂Cl₂ were added batchwise CaO (6.0 mmol) and NsONHCO₂R (3.0 mmol) at 0 °C. The reaction was monitored by TLC until completion (see Table 1), and then the crude mixture was diluted with 10 mL of CH₂Cl₂ and filtered. After solvent evaporation under reduced pressure, the residue was purified by flash chromatography on silica gel (70:30 hexane/ethyl acetate).

5a: mp 133–134 °C (pentane/CHCl₃); [α]_D +72.3 (c 4.8, CHCl₃); IR 1767, 1733, 1723 cm⁻¹; ¹H NMR δ 0.52 (s, 3 H), 0.79 (s, 3 H), 0.84 (s, 3 H), 0.94–2.58 (m, 9 H), 1.22 (t, *J* = 7.2 Hz, 3 H), 1.97 (s, 3 H), 2.29 (s, 3 H), 3.76 (d, *J* = 7.2 Hz, 1 H), 4.11–4.31 (m, 2 H), 4.29–4.36 (m, 1 H), 5.28 (d, *J* = 7.2 Hz, 1 H), 5.56–5.64 (br, 1 H), 6.82 (s, 1 H), 7.02–7.52 (m, 6 H); ¹³C NMR δ 11.1, 13.9, 14.2, 19.2, 20.6, 21.0, 27.9, 31.6, 32.7, 47.2, 48.2, 50.1, 50.7, 51.9, 62.9, 67.0, 81.8, 128.1, 129.5, 132.5, 136.6, 138.0, 157.7, 162.6, 200.5; ESI MS *m/z* 609 (M⁺ + 1). Anal. Calcd for C₃₃H₄₀N₂O₇S: C, 65.11; H, 6.62; N, 4.60; S, 5.27. Found: C, 65.18; H, 6.69; N, 4.62; S, 5.32.

6a: mp 164–165 °C (pentane/CHCl₃); [α]_D +100.8 (c 4.9, CHCl₃); IR 1766, 1720 cm⁻¹; ¹H NMR δ 0.53 (s, 3 H), 0.79 (s, 3 H), 0.82 (s, 3 H), 1.00–2.58 (m, 9 H), 1.42 (s, 9 H), 1.98 (s, 3 H), 2.29 (s, 3 H), 3.79 (d, *J* = 7.2 Hz, 1 H), 4.22–4.31 (m, 1 H), 5.28 (d, *J* = 7.2 Hz, 1 H), 5.52–5.68 (br, 1 H), 6.83 (s, 1 H), 7.26–7.55 (m, 6 H); ¹³C NMR δ 10.9, 13.9, 19.1, 20.7, 21.1, 27.7, 28.0, 31.6, 32.9, 47.2, 48.2, 50.1, 51.1, 52.1, 67.0, 81.8, 82.0, 128.2, 129.5, 132.5, 136.8, 137.9, 156.3, 162.9, 200.9; ESI MS *m/z* 637 (M⁺ + 1). Anal. Calcd for C₃₅H₄₄N₂O₇S: C, 66.01; H, 6.96; N, 4.40; S, 5.03. Found: C, 66.12; H, 6.98; N, 4.46; S, 5.10.

5b: viscous oil; [α]_D -44.8 (c 5.2, CHCl₃); IR 1769, 1738 cm⁻¹; ¹H NMR δ 0.85 (d, *J* = 6.3 Hz, 3 H), 0.95–1.78 (m, 15 H), 1.84–2.55 (m, 4 H), 2.20–2.38 (m, 2 H), 3.01–3.08 (m, 1 H), 3.95–4.42 (m, 2 H), 4.95 (dt, *J*_d = 4.5 Hz, *J*_t = 10.5 Hz, 1 H), 7.09–7.20 (m, 1 H), 7.25–7.36 (m, 4 H); ¹³C NMR δ 14.2, 19.6, 21.7, 26.6, 26.9, 29.7, 31.4, 32.9, 34.4, 40.0, 41.5, 49.8, 49.9, 63.2, 76.6, 125.1, 125.5, 127.8, 150.9, 157.4, 162.7, 200.1; ESI MS *m/z* 428 (M⁺ + 1). Anal. Calcd for C₂₅H₃₃NO₅: C, 70.23; H, 7.78; N, 3.26. Found: C, 70.34; H, 7.56; N, 3.30.

6b: mp 58–59 °C (pentane); [α]_D -10.6 (c 5.1, CHCl₃); IR 1762, 1724 cm⁻¹; ¹H NMR δ 0.84 (d, *J* = 6.6 Hz, 3 H), 0.97–1.81 (m, 12 H), 1.42 (s, 9 H), 1.83–2.28 (m, 4 H), 2.27–2.46 (m, 2 H), 3.05–3.23 (m, 1 H), 4.94 (dt, *J*_d = 4.5 Hz, *J*_t = 10.5 Hz, 1 H),

7.05–7.20 (m, 1 H), 7.21–7.38 (m, 4 H); ¹³C NMR δ 19.3, 21.7, 26.4, 27.0, 27.5, 27.9, 31.4, 33.1, 34.3, 40.1, 41.5, 49.6, 50.1, 50.4, 76.5, 82.8, 125.1, 125.6, 127.7, 150.7, 156.0, 163.2, 200.1; ESI MS *m/z* 456 (M⁺ + 1). Anal. Calcd for C₂₇H₃₇NO₅: C, 71.18; H, 8.19; N, 3.07. Found: C, 71.14; H, 8.11; N, 3.03.

7a: mp 167–168 °C (pentane/CHCl₃); [α]_D -22.1 (c 2.9, CHCl₃); IR 1750, 1736, 1721 cm⁻¹; ¹H NMR δ 0.52 (s, 3 H), 0.69 (s, 3 H), 0.83 (s, 3 H), 0.98–2.45 (m, 11 H), 1.25 (t, *J* = 7.2 Hz, 3 H), 2.02 (s, 3 H), 2.29 (s, 3 H), 3.81 (d, *J* = 7.2 Hz, 1 H), 3.89–3.92 (m, 1 H), 4.18 (q, *J* = 7.2 Hz, 2 H), 5.35 (d, *J* = 7.2 Hz, 1 H), 5.64–5.73 (br, 1 H), 6.85 (s, 1 H), 7.10–7.53 (m, 6 H); ¹³C NMR δ 11.2, 14.1, 17.3, 20.5, 21.0, 21.3, 22.1, 28.0, 32.0, 38.5, 45.2, 46.8, 47.2, 48.4, 50.3, 62.8, 67.0, 82.9, 128.0, 128.7, 129.6, 132.7, 136.3, 138.2, 158.1, 164.8, 196.9; ESI MS *m/z* 623 (M⁺ + 1). Anal. Calcd for C₃₄H₄₂N₂O₇S: C, 65.57; H, 6.80; N, 4.50; S, 5.15. Found: C, 65.63; H, 6.85; N, 4.52; S, 5.22.

8a: mp 206–207 °C (pentane/CHCl₃); [α]_D -27.1 (c 2.1, CHCl₃); IR 1740, 1718, 1708 cm⁻¹; ¹H NMR δ 0.52 (s, 3 H), 0.72 (s, 3 H), 0.84 (s, 3 H), 0.98–2.45 (m, 11 H), 1.41 (s, 9 H), 2.02 (s, 3 H), 2.24 (s, 3 H), 3.51–3.64 (m, 1 H), 3.84 (d, *J* = 7.2 Hz, 1 H), 5.29 (d, *J* = 7.2 Hz, 1 H), 5.63–5.85 (br, 1 H), 6.82 (s, 1 H), 7.10–7.62 (m, 6 H); ¹³C NMR δ 11.2, 17.2, 20.7, 21.0, 21.2, 21.3, 27.7, 28.1, 31.6, 37.9, 44.8, 47.2, 47.7, 48.2, 50.3, 66.5, 82.0, 82.5, 128.0, 128.5, 129.4, 132.4, 136.7, 138.1, 156.5, 165.2, 197.0; ESI MS *m/z* 651 (M⁺ + 1). Anal. Calcd for C₃₆H₄₆N₂O₇S: C, 66.44; H, 7.12; N, 4.30; S, 4.93. Found: C, 66.49; H, 7.15; N, 4.38; S, 5.02.

9b: bp 169–170 °C (7 mmHg); [α]_D +27.4 (c 4.6, CHCl₃); IR 1747, 1735, 1723 cm⁻¹; ¹H NMR δ 0.75–1.11 (m, 4 H), 0.84 (d, *J* = 6.5 Hz, 3 H), 1.18 (s, 3 H), 1.24 (t, *J* = 7.3 Hz, 3 H), 1.30 (s, 3 H), 1.38–1.48 (m, 1 H), 1.55–1.66 (m, 2 H), 1.82–1.87 (m, 1 H, CH), 2.08 (d, *J* = 1.3 Hz, 1 H), 2.11 (d, *J* = 1.3 Hz, 1 H), 2.30 (s, 3 H), 4.04–4.19 (m, 2 H), 4.94 (dt, *J*_d = 4.5 Hz, *J*_t = 10.7 Hz, 1 H), 7.10–7.15 (m, 1 H), 7.22–7.27 (m, 4 H); ¹³C NMR δ 14.1, 21.6, 25.6, 26.6, 27.4, 28.3, 31.3, 34.2, 36.3, 39.7, 41.3, 49.5, 50.1, 62.9, 77.3, 125.3, 128.0, 151.2, 158.9, 164.9, 199.0; GC MS *m/z* 415 (M⁺, 12), 201 (70), 157 (14), 156 (23), 129 (100), 128 (32), 119 (42), 111 (59), 110 (10), 105 (48), 91 (37), 84 (21), 83 (21), 55 (10), 43 (13), 41 (13). Anal. Calcd for C₂₄H₃₃NO₅: C, 69.37; H, 8.00; N, 3.37. Found: C, 69.41; H, 8.07; N, 3.40.

10b: bp 193–194 °C (7 mmHg); [α]_D +15.7 (c 4.2, CHCl₃); IR 1736, 1722 cm⁻¹; ¹H NMR δ 0.72–1.75 (m, 6 H), 0.85 (d, *J* = 6.6 Hz, 3 H), 1.24 (s, 3 H), 1.34 (s, 3 H), 1.46 (s, 9 H), 1.87–2.08 (m, 2 H), 2.24 (d, *J* = 2.2 Hz, 1 H), 2.26 (d, *J* = 2.2 Hz, 1 H), 2.34 (s, 3 H), 4.95 (dt, *J*_d = 4.4 Hz, *J*_t = 11.0 Hz, 1 H), 7.10–7.20 (m, 1 H), 7.24–7.31 (m, 4 H); ¹³C NMR δ 21.6, 26.3, 26.8, 27.2, 27.9, 28.6, 31.3, 34.3, 37.0, 40.0, 41.3, 49.6, 50.2, 77.6, 82.4, 125.3, 125.5, 128.1, 150.8, 157.5, 165.3, 199.2; GC MS *m/z* 342 (M⁺ - 101, <1), 128 (88), 120 (10), 119 (100), 118 (55), 109 (57), 91 (33), 41 (10). Anal. Calcd for C₂₆H₃₇NO₅: C, 70.40; H, 8.41; N, 3.16. Found: C, 70.33; H, 8.34; N, 3.11.

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